

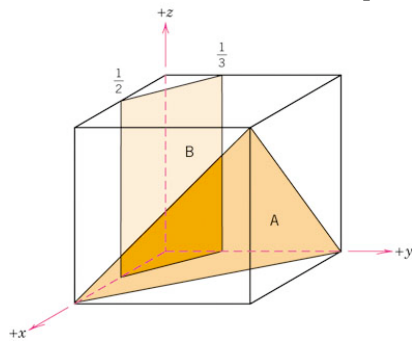
HW #10

1. Sketch a unit cell of the following crystals:

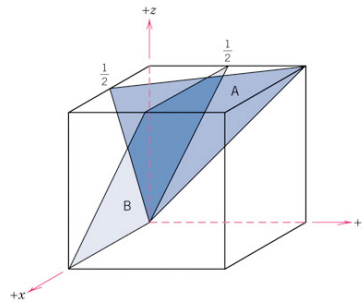
- a) Copper, which has a BCC structure with base $(0,0,0)$
- b) Gallium Arsenide (GaAs), which has an FCC structure with Ga atoms at $(0,0,0)$, and As atoms at $(\frac{1}{4}, \frac{1}{4}, \frac{1}{4})$
- c) Cesium chloride (CsCl) which has a simple cubic structure with Cs atoms at $(0,0,0)$, and Cl atoms at $(\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$.

2. Determine the Miller indices for the planes shown in the following unit cells:

i)



(ii)



3. **(Derivation for Exam 2)** Using geometrical arguments, derive Bragg's law for a simple cubic system. Make sure to include a drawing with a source, detector, incoming and reflected rays, and appropriate labels for lengths and angles.