

Kasap example 4.7

a) $E_{center} \approx 5 \text{ eV}$, then

$$g(E_{center}) = 8\pi\sqrt{2} \left(\frac{m_e}{h^2} \right)^{3/2} E_{center}^{1/2} = 8\pi\sqrt{2} \left[\frac{9.11 \times 10^{-31} \text{ kg}}{(6.626 \times 10^{-34} \text{ Js})^2} \right]^{3/2} \left(5 \text{ eV} \times \frac{1.6 \times 10^{-19} \text{ J}}{\text{eV}} \right)^{1/2} =$$

$$= 9.5 \times 10^{46} / \text{Jm}^3 \times \frac{1.6 \times 10^{-19} \text{ J}}{\text{eV}} \times \frac{1 \text{ m}^3}{10^6 \text{ cm}^3} = 1.52 \times 10^{22} / \text{eV cm}^3$$

b) $kT \approx 0.026 \text{ eV}$ @ $T = 300 \text{ K}$

$$E_{kt} = E_{center} - kT = 5 \text{ eV} - 0.026 \text{ eV} = 4.974 \text{ eV}$$

$$g(E_{kt}) = 8\pi\sqrt{2} \left[\frac{9.11 \times 10^{-31} \text{ kg}}{(6.626 \times 10^{-34} \text{ Js})^2} \right]^{3/2} \left(4.974 \text{ eV} \times \frac{1.6 \times 10^{-19} \text{ J}}{1 \text{ eV}} \right)^{1/2} =$$

$$= 9.48 \times 10^{46} / \text{Jm}^3 = 1.516 \times 10^{22} / \text{eV cm}^3$$

since kT is small compared to E_{center} , we see that the densities of states are similar, so we can approximate the number of states by

$$g(E) \cdot \delta E = (1.52 \times 10^{22} / \text{eV cm}^3) (0.026 \text{ eV}) = 3.94 \times 10^{20} \text{ states/cm}^3$$

or, doing the exact calculation:

$$\int_{E_{center}-kT}^{E_{center}} g(E) dE = 8\pi\sqrt{2} \left(\frac{m_e}{h^2} \right)^{3/2} \frac{2}{3} E^{3/2} \Big|_{E_{center}-kT}^{E_{center}} =$$

$$= 8\pi\sqrt{2} \left[\frac{9.11 \times 10^{-31} \text{ kg}}{(6.626 \times 10^{-34} \text{ Js})^2} \right]^{3/2} \cdot \frac{2}{3} \left[\left(5 \text{ eV} \times \frac{1.6 \times 10^{-19} \text{ J}}{\text{eV}} \right)^{3/2} - \left(4.974 \text{ eV} \times \frac{1.6 \times 10^{-19} \text{ J}}{\text{eV}} \right)^{3/2} \right]$$

$$= 3.95 \times 10^{26} / \text{m}^3 \times \frac{1 \text{ m}^3}{10^6 \text{ cm}^3} = 3.95 \times 10^{20} / \text{cm}^3$$

which is very similar to the answer above

c)

$$g(E=0.026 \text{ eV}) = 8\pi\sqrt{2} \left[\frac{9.11 \times 10^{-31} \text{ kg}}{(6.626 \times 10^{-34} \text{ Js})^2} \right]^{3/2} \left(0.026 \text{ eV} \times \frac{1.6 \times 10^{-19} \text{ J}}{\text{eV}} \right)^{1/2} =$$
$$= 6.84 \times 10^{45} / \text{Jm}^3 = 1.1 \times 10^{21} / \text{eV cm}^3$$

d) the number of states per unit volume:

$$g(E=0.026 \text{ eV}) \cdot \delta E = (1.1 \times 10^{21} / \text{eV cm}^3) (0.026 \text{ eV}) = 2.8 \times 10^{19} / \text{cm}^3$$